

Studies on the Synthesis of the Antitumor Agents Esperamicin A₁/Calicheamicin γ_1 : Stereospecific Synthesis of the 12 β -Hydroxybicyclo[7.3.1] Diynene Core Structure via a Co₂(CO)₈-Acetylene-Mediated Synclinal Aldol Reaction

Philip Magnus,* Hirokazo Annoura, and John Harling

Department of Chemistry, University of Texas at Austin, Austin, Texas 78712

Received December 5, 1989

Summary: Treatment of **11** with *n*-Bu₂BOTf/amine bases resulted in its stereospecific conversion into the 12 β -hydroxybicyclo[7.3.1] diynene-CO₂(CO)₆ adduct **16**.

The key reaction in our approach to the synthesis of the potent antitumor compounds esperamicin/calicheamicin **1**¹ involves the use of the Pettit-Nicholas Co₂(CO)₈-propargyl cation chemistry to convert **2** into **3** (Scheme I).² While we have been able to introduce all of the functionality required for calicheamicin (C-1,2 double bond, C-2 nitrogen, C-3 oxygen, C-13 allylic trisulfide, and C-12 oxygen) into the cobalt-decomplexed substrate **4**, despite extensive experimentation we have not been able to successfully introduce the C-12 hydroxyl group by carrying out the Pettit-Nicholas reaction at the aldehyde oxidation level.³ Scheme II shows a number of unsuccessful attempts to close the C-1, C-12 bond with a functional group at C-12. Even treatment of **9** with Me₃SiOTf/2,6-di-*tert*-butyl-4-methylpyridine (Pummerer reaction) only gave **3** (65%). Presumably, the PhS(O) group is activated by silylation (on oxygen) and then ionizes to the stabilized cation **2a** (PhSOSiMe₃ leaving group). The failure of the derivatives **5**-**9** to give **10** may be attributed to the lack of reactivity of **9a** (too stabilized). In support of this the keto aldehyde **11** was the major isolated product in all the attempts to convert **5**-**9** into **10**. Sulfonium and oxonium ions should be aligned antiperiplanar to the enol π -system, and this appears to be difficult to achieve (steric).⁴

Furthermore, the product **10** may be unstable in respect to further ionization. Consequently, it was reasoned that a mild synclinal aldol reaction mediated by a boron enolate derivative might allow access to derivatives of **10**.⁵ It should be noted that the other synthetic approaches to the esperamicins/calicheamicins arrive at 12-hydroxy bicyclo[7.3.1] diynene derivatives as a mixture of epimers at C-12.¹ Treatment of **12** with propargyl alcohol/Pd(Ph₃P)₄ (cat.)/CuI/*n*-BuNH₂ gave **13** (80%),⁶ which was oxidized using Swern conditions to the unstable aldehyde **14** (Scheme III). Cobalt complexation gave **15** (92%)⁷ and removal of the MEM protecting group using Me₂BBr-(THF/CH₂Cl₂)⁸ gave the keto aldehyde **11** (73%). The yield of **11** was considerably decreased if THF was not added in this deprotection (30-35% of **11**). When **11** was treated with *n*-Bu₂BOTf/DABCO/NEt₃/CH₂Cl₂-THF the aldol product **16** was isolated as a single stereoisomer [45% after chromatography over silica gel (twice)] (Scheme IV). The 9-H to 12 α -H proton-proton coupling is 1.59 Hz, which corresponds very closely to that observed in the natural products **1** (1.8 Hz). Surprisingly, when **16** was treated with I₂/THF/room temperature the aromatized diiodide **18** was isolated in 60% yield. This should be contrasted with **3**, which gave **4** (70%) with only small amounts of the iodide **19**. In contrast, when **16** was treated with *N*-methylmorpholine *N*-oxide/THF, *t*-BuOH/room temperature, the nonaromatized diynene **17** was isolated in 76% yield. Exposure of **17** to I₂/THF/room temperature did not give **18**. A plausible explanation for the aromatization of **16** to **18** might involve the hypoiodide **20**, which can deliver I[•] to the adjacent triple bond to give **21** and initiate cycloaromatization to **19**. The above route to **17** was made more convergent by utilizing the modifications described by Kadow⁹ in his adaptation of our original sequence.² Treatment of **22** with acetylene carboxaldehyde diethyl acetal under the usual coupling conditions gave **23** (68%) (Scheme V). Removal of the trimethylsilyl group with aqueous lithium hydroxide and treatment of the terminal acetylene with lithium bis(trimethylsilyl)amide generated the lithio derivative **24**, which was quenched with **25** to give **26** which undergoes silyl migration pro-

(1) The structures of the esperamicins and calicheamicins were reported in 1987. Golik, J.; Dubay, G.; Groenewold, G.; Kawaguchi, H.; Konishi, M.; Krishnan, B.; Ohkuma, H.; Saitoh, K.; Doyle, T. W. *J. Am. Chem. Soc.* **1987**, *109*, 3462. Lee, M. D.; Dunne, T. S.; Siegel, M. M.; Chang, C. C.; Morton, G. O.; Borders, D. B. *J. Am. Chem. Soc.* **1987**, *109*, 3464. Lee, M. M.; Dunne, T. S.; Chang, C. C.; Ellestad, G. A.; Siegel, M. M.; Morton, G. O.; McGahren, W. G.; Borders, D. B. *J. Am. Chem. Soc.* **1987**, *109*, 3466. For other synthetic approaches, see: Danishefsky, S. J.; Mantlo, N. B.; Yamashita, D. S.; Schultz, G. *J. Am. Chem. Soc.* **1988**, *110*, 6980. Schreiber, S. L.; Kiessling, L. L. *Tetrahedron Lett.* **1989**, *30*, 433. Kende, A. S.; Smith, C. A. *Tetrahedron Lett.* **1988**, *29*, 4217. Nicolaou, K. C.; Zuccarello, G.; Ogawa, W.; Schweiger, E. J.; Kumazawa, T. *J. Am. Chem. Soc.* **1988**, *110*, 4866. Nicolaou, K. C.; Ogawa, Y.; Zuccarello, G.; Kataoka, H. *J. Am. Chem. Soc.* **1988**, *110*, 7247. Tomioka, K.; Fujita, H.; Koga, K. *Tetrahedron Lett.* **1989**, *30*, 851. Schoenen, F. G.; Porco, J. A. Jr.; Schreiber, S. L.; VanDuynne, G. D.; Clardy, J. *Tetrahedron Lett.* **1989**, *30*, 3765. Haseltine, J. N.; Danishefsky, S. J.; Schulte, G. *J. Am. Chem. Soc.* **1989**, *111*, 7638.

(2) Magnus, P.; Carter, P. A. *J. Am. Chem. Soc.* **1988**, *110*, 1626. Magnus, P.; Lewis, R.; Huffman, J. C. *J. Am. Chem. Soc.* **1988**, *110*, 6921. Magnus, P.; Lewis, R. *Tetrahedron Lett.* **1989**, *30*, 1905. The construction of the trisulfide functionality, see: Magnus, P.; Lewis, R.; Bennett, F. *J. Chem. Soc., Chem. Commun.* **1989**, 916. Introduction of the β -hydroxyl and 1,2-double bond, see: Magnus, P.; Bennett, F. *Tetrahedron Lett.* **1989**, 3637. Introduction of the C-2 nitrogen and C-12 β oxygen, see: Magnus, P.; Lewis, R., unpublished results.

(3) For very recent examples of the stereoselective aldol reaction of Co₂(CO)₈-acetylene aldehyde complexes, see: Ju, J.; Reddy, B. B.; Khan, M.; Nicholas, K. M. *J. Org. Chem.* **1989**, *54*, 5426. Mukai, C.; Nagami, K.; Hanaoka, M. *Tetrahedron Lett.* **1989**, *30*, 5623 and 5627. For a comprehensive review of Co₂(CO)₈-complexed propargyl cations, see: Nicholas, K. M. *Acc. Chem. Res.* **1987**, *20*, 208. For the original report of the Co₂(CO)₈-propargyl cation chemistry, see: Nicholas, K. M.; Pettit, R. *Tetrahedron Lett.* **1971**, 3475.

(4) The newly formed C-1/C-2 bond in **3/16** is axial and the 12 β -substituent (corresponding to an antiperiplanar aldol transition state) is equatorial. There is poor overlap between C-1 and C-12 in this orientation.

(5) Evans, D. A.; Vogel, E.; Nelson, J. V. *J. Am. Chem. Soc.* **1979**, *101*, 6120. Mukaiyama, T.; Inoue, T. *Chem. Lett.* **1976**, 559. Fenzl, W.; Koster, R. *Justus Liebig's Ann. Chem.* **1975**, 1322. Masamune, S.; Mori, S.; Van Horn, D.; Brooks, D. W. *Tetrahedron Lett.* **1979**, 6120. Paterson, I.; Lister, M. A. *Tetrahedron Lett.* **1988**, *29*, 585 and references therein pertaining to chiral boron reagents. For a comprehensive review on the aldol reaction, see: Heathcock, C. H. In *Asymmetric Synthesis*; Morrison, J. D., Ed.; Academic Press: New York, 1984; Vol 3, pp 111-212. For a recent description of transition state geometry in the aldol reaction, see: Denmark, S. E.; Henke, B. R. *J. Am. Chem. Soc.* **1989**, *111*, 8032.

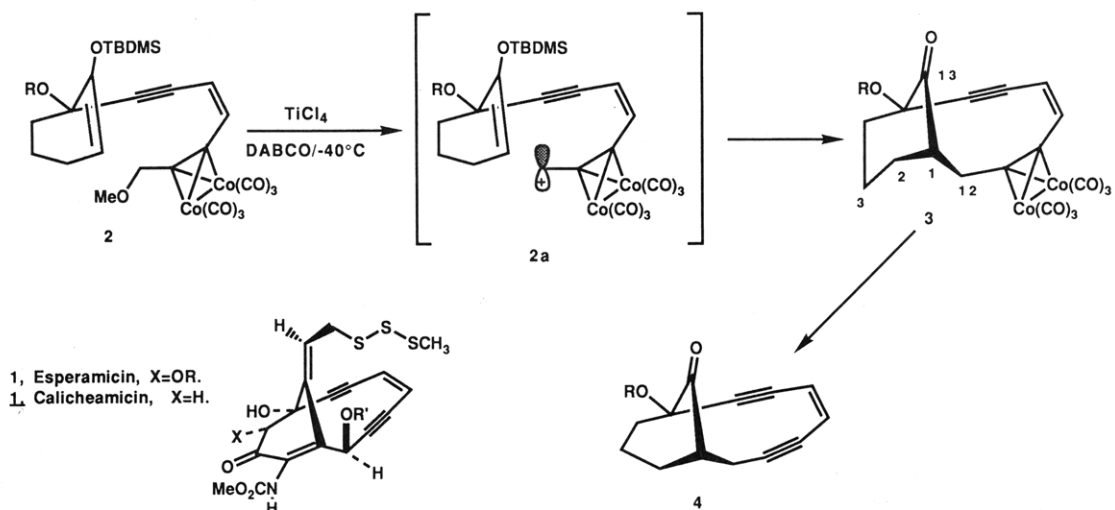
(6) Ratovelomanana, V.; Linstrumelle, G. *Tetrahedron Lett.* **1984**, *25*, 6001. Stephens, R. D.; Casrto, C. E. *J. Org. Chem.* **1963**, *28*, 3313.

(7) The less sterically hindered triple bond is preferentially complexed with Co₂(CO)₈. A small amount (ca. 5%) of the other regioisomer is formed, along with <1% of the bis adduct. The desired adduct **15** is purified by chromatography over silica gel prior to use.

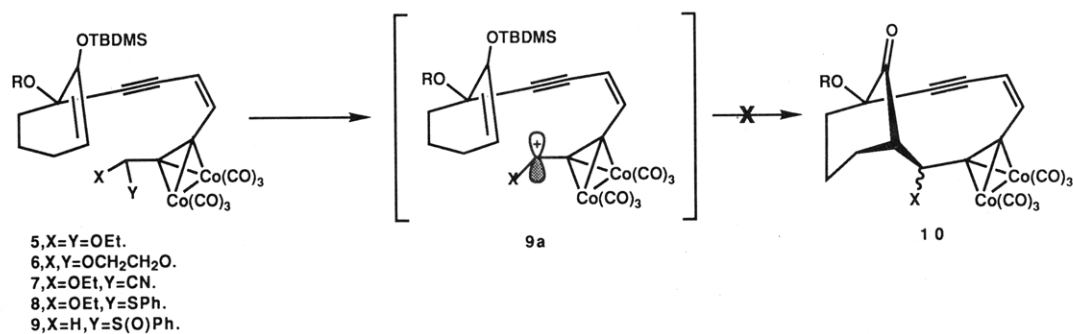
(8) Guindon, Y.; Morton, H. E.; Yoakim, C. *Tetrahedron Lett.* **1983**, *24*, 3969; *J. Org. Chem.* **1984**, *49*, 3912.

(9) Kadow, J. F.; Saulnier, M. G.; Tun, M. M.; Langley, D. R.; Vyas, D. M. *Tetrahedron Lett.* **1989**, *30*, 3499.

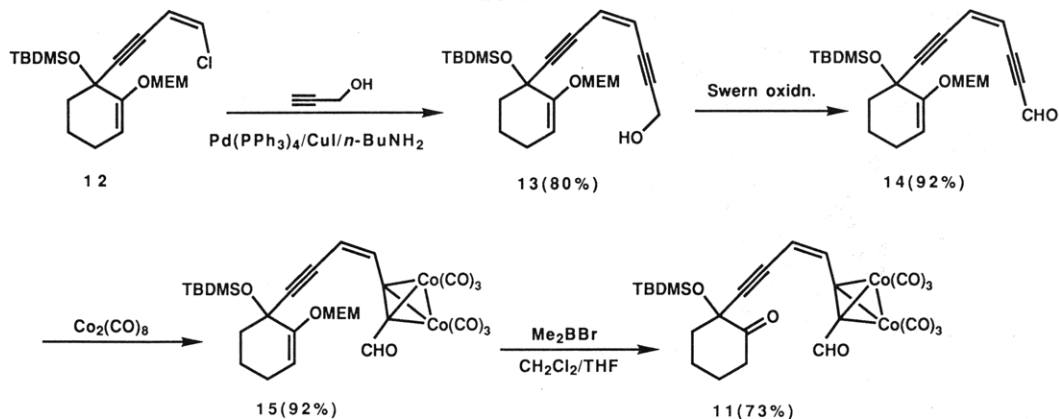
Scheme I (R = TBDMS)



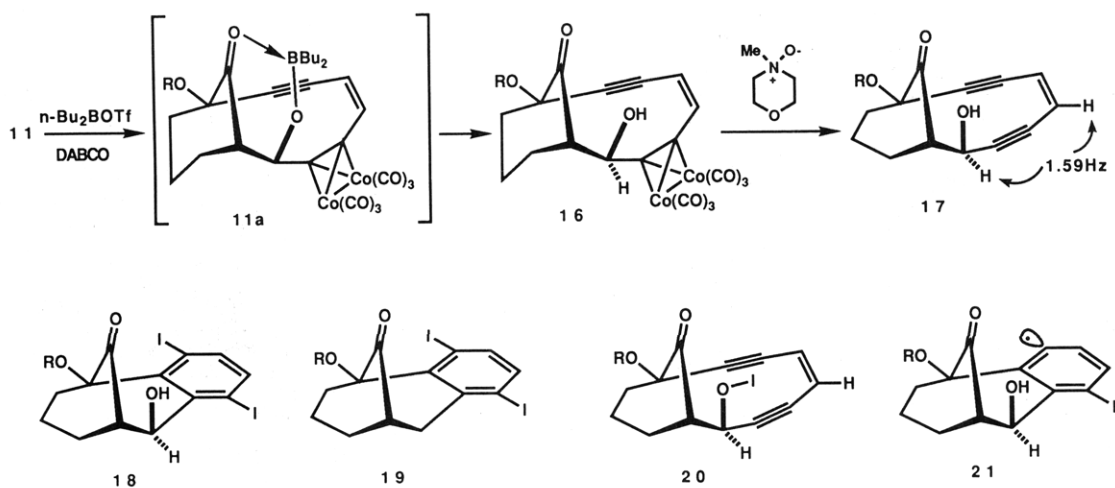
Scheme II



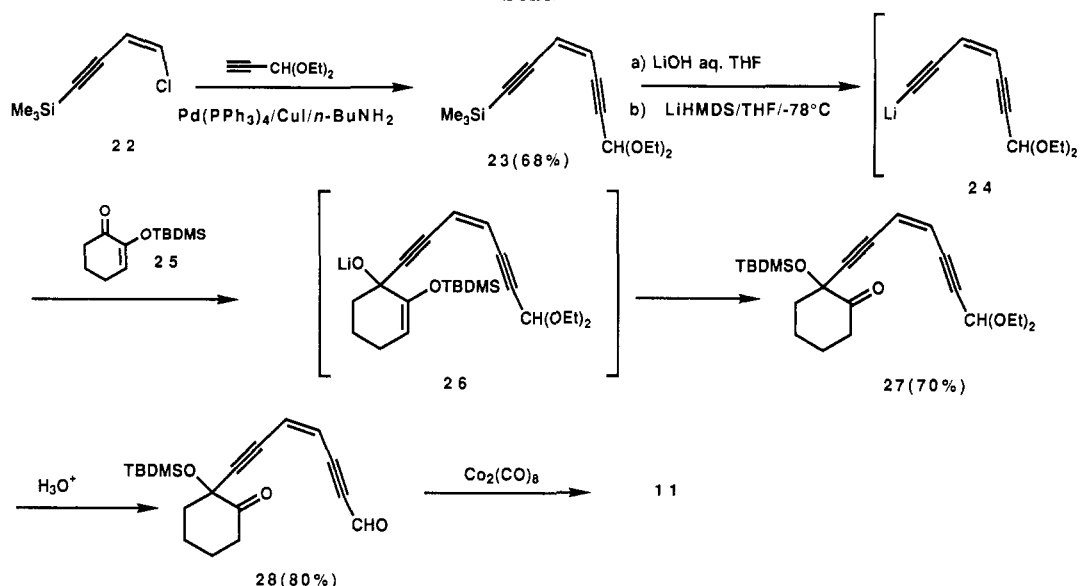
Scheme III



Scheme IV



Scheme V



viding 27 (70%). Hydrolysis of 27 with trifluoroacetic acid/chloroform gave 28 (80%) which was converted directly into 11 (90%) by treatment with dicobalt octacarbonyl. This route allows access to 12 β -hydroxybicyclo[7.3.1] diynene 17 in eight steps from *cis*-dichloroethylene and trimethylacetylene in an overall yield of 14% (unoptimized). The synclinal aldol mediated stereospecific synthesis of the 12 β -hydroxybicyclo[7.3.1] diynene system 17 should allow the examination of bridgehead enol

chemistry (C-1, C-13) in the presence of the 12 β -substituent.

Acknowledgment. The National Institutes of Health are thanked for their financial support of this work.

Supplementary Material Available: Spectral data for compounds 11 and 13-18 (1 page). Ordering information is given on any current masthead page.

Articles

1-Hydroxy-3-amino-2-piperidone (δ -N-Hydroxycycloornithine) Derivatives: Key Intermediates for the Synthesis of Hydroxamate-Based Siderophores

Teodozj Kolasa and Marvin J. Miller*

Department of Chemistry, University of Notre Dame, Notre Dame, Indiana 46556

Received September 19, 1989

Several routes for the synthesis of δ -N-(benzyloxy)cycloornithine (2) from glutamic acid derived starting materials are described. Efficient methods were developed for the synthesis of glutamic acid γ -semialdehyde and δ -hydroxynorvaline derivatives as key substrates for preparation of δ -N-hydroxycycloornithine analogues. Thus, the best approaches to the synthesis of 2 were: (1) reductive cyclization of an *N*-hydroxysuccinimide ester of the *O*-benzyloxime 4 of α -amino-protected glutamic acid γ -semialdehyde 5 and (2) cyclization of the *N*-(benzyloxy)amide of δ -bromonorvaline (7).

Introduction

The pseudomonads represent a diversified group of Gram-negative bacteria widely distributed in the soil. The fluorescent pseudomonads which belong to group 1, according to their genetic homology, release yellow-green fluorescent pigments when grown under iron-deficient conditions. These pigments are the siderophores of the fluorescent pseudomonads and are called pyoverdines¹⁻⁴

or pseudobactins⁵⁻⁹ and serve as a biological source of iron for these bacteria.

(1) Wendenbaum, S.; Demange, P.; Dell, A.; Meyer, J. M.; Abdallah, M. A. *Tetrahedron Lett.* **1983**, *24*, 4877.
(2) Poppe, K.; Taraz, K.; Budzikiewicz, H. *Tetrahedron* **1987**, *43*, 2261.

(3) Briskot, G.; Taraz, T.; Budzikiewicz, H. *Z. Naturforsch.* **1986**, *41c*, 497.

(4) Cody, Y. S.; Gross, D. C. *J. Bacteriol.* **1987**, *169*, 2207.

(5) Van der Hofstad, G. A.; Marug, J. D.; Verjans, G. M.; Weisbeck, P. J. Iron, siderophores, and plant diseases, NATO ASI Series; Swinburne, T. R., Ed.; Plenum Press: New York, 1986; Vol. 117, pp 71-75.

(6) Yang, C. C.; Leong, J. *Biochemistry* **1984**, *23*, 3534.

(7) Buyer, J. S.; Wright, J. M.; Leong, H. *Biochemistry* **1986**, *25*, 5492.

(8) Teintze, M.; Hossain, M. B.; Barnes, C. L.; Leong, J.; Van der Helm, D. *Biochemistry* **1981**, *20*, 6446.